# Shivaji University, Kolhapur



Syllabus For

B. Sc. Part-I Nanotechnology Under CBCS

to be implemented from the academic year 2018-19

(June 2018) onwards.

# SHIVAJI UNIVERSITY, KOLHAPUR

# (B. Sc. Nanotechnology course)

# F. Y. B. Sc., Sem. I

### **Course Structure**

| Course No | Title                           | Lecture | Practical | Examination/<br>Evaluation of<br>marks/Semes<br>ter |
|-----------|---------------------------------|---------|-----------|---|
| DSC-37A   | Introduction to nanoscience and | 37      | -         |   |
|           | Nanotechnology                  |         |           | 50  |
| DSC-38A   | Physico-Chemical aspects of     | 38      | -         | 50  |
|           | Nanoscience and                 |         |           |   |
|           | Nanotechnology                  |         |           |   |
| Practical | Laboratory Course-I             | -       | 60        | -   |
| DSC-37A   |                                 |         |           |   |
| and DSC-  |                                 |         |           |   |
| 38A       |                                 |         |           |   |

### (B. Sc. Nanotechnology) F. Y. B. Sc., Sem. I Syllabus Paper I

### Title of Paper: Introduction to Nanoscience and Nanotechnology

#### Subject Code: DSC-37A

#### Unit I: Fundamentals of Nanoscience and Nanotechnology

Definitions, Relationship and Differences. Nano and Nature: Nanoscopic Colours (Butterfly Wings), Bioluminescence (Fireflies), Tribiology (Geckos sticky feet, lotus leaf effect). Introduction to hydrophilic and hydrophobic materials. Nanotechnology timeline, Pre-18<sup>th</sup> Century, 19<sup>th</sup> Century, 20<sup>th</sup> Century and 21<sup>st</sup> Century. Future perspectives of nanoscience and nanotechnology.

#### **Unit II: Nanoscale Science**

Interconversion of units. Introduction to surface area to volume ratio and aspect ratio. Difference between surface area to volume ratio of bulk materials and nanomaterials (sphere, hollow sphere, rods, hollow rods, cubes and hollow cubes) and related numerical problems. Difference in aspect ratio of bulk wire and nanowire and related numerical problems. Nanomaterials and wavelength of light.

### **Unit III: Classification of Nanomaterials**

Introduction to dimensional growth process. Classification of nanomaterials into 0D, 1D, 2D and 3D. Relationship between dimension and shape of nanomaterials (Quantum dots, Quantum wires, Carbon nanotubes, Bucky balls, Fullerenes). Introduction to size effect on electronic and optical properties (Quantum confinement).

### Unit IV: Introduction to Self-assembled Biological Nanomaterials in Nature 8 L

Fundamentals of nanoscale self-assembly process involved in important functional biomolecules such as Nucleic acid (DNA and RNA), Proteins, Enzymes. Cell structure and organelles, nanoscale assembly of cellular components (cell membrane and liposomes). Nanoscale assembly of microorganisms (virus).

#### **References:**

1. Introduction to nanoscience and nanotechnology, CRC Press, Tylor and Francis Group, Boca Raton, G. L. Hornyak, H. F. Tibbals, J. Dutta and J J. Moore.

2. Introductory Nanoscience: Physical and Chemical Concepts, CRC Press, Tylor and Francis Group, Boca Raton, M. Kuno.

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### (B. Sc. Nanotechnology) F. Y. B. Sc., Sem. I Syllabus Paper II

Title of Paper: Physico-Chemical, Biologicalaspects of Nanoscience and Nanotechnology Subject Code: DSC-38A

### Unit I:Fundamentals of Atomic Structure and Bonding

Bohr's atomic structure, Bohr's atomic radii, comparative size of nanomaterials and atomic size, electronic configuration, energy levels of shells and related numerical problems on excitation of electrons from lower to higher energy level. Concept of quantization of energy. Arrangement of atoms in solids (two dimension crystal structures and three dimension crystal structure). Bonding in solids (MOT), bonding and antibonding states. Electronic structures of solids.

### **Unit II:Crystal Structure**

Lattices, basis of crystallographic planes and direction. Simple bcc, and fcc crystal structures.

### Unit III: Types of Solid and Phase Diagram

Single Phase alloys, Semiconductors, insulators and oxide materials. Basic terms involved in phase diagram: system, surrounding, component, co-ordinates, phase equilibrium, phase diagram. Lever rule, Gibb's phase rule, phase diagram of Pb-Sn system.

### Unit IV: Terminology and Techniques in Nanobiotechnology

Definitions, Scopes and applications of Biotechnology, Nanobiotechnology, Biomolecular Nanotechnology, Biomedical Nanotechnology, Green Nanotechnology. Fundamentals and introduction to techniques such as mechanical extraction, physical methods of homogenization, centrifugation, dialysis, electrophoresis and chromatography techniques for purification of biomolecules and microsopy.

#### **References:**

- 1. Materials Science and Engineering –V. raghavan
- 2. Elements of Material Science and Engineering-H. Vanvlach (4<sup>th</sup> Edition)
- 3. Nanotechnology-S. K. Kulkarni (3<sup>rd</sup> Edition)

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# (B. Sc. Nanotechnology course) F. Y. B. Sc., Sem. II

### **Course Structure**

| Course No | Title                           | Lecture | Practical | Examination/<br>Evaluation of<br>marks/Semes<br>ter |
|-----------|---------------------------------|---------|-----------|---|
| DSC-37B   | Techniques for the synthesis of | 37      | -         |   |
|           | Nanomaterials                   |         |           | 50  |
| DSC-38B   | Basic Characterization          | 38      | -         | 50  |
|           | Techniques for Nanomaterials    |         |           |   |
| Practical | Laboratory Course-I             | -       | 60        | 50  |
| DSC-37B   |                                 |         |           |   |
| and DSC-  |                                 |         |           |   |
| 38B       |                                 |         |           |   |

### (B. Sc. Nanotechnology) F. Y. B. Sc., Sem. II Syllabus Paper I

Title of Paper: Techniques for the synthesis of Nanomaterials

### Subject Code: DSC-37B

#### **Unit I:Basics of Fabrication Methods**

Top-Down fabrication methods –Types of Top-Down fabrication methods (mechanosynthesis, thermal, high energy, chemical fabrication and lithography-concepts with examples only). Bottom-Up fabrication methods-Types of Bottom-Up fabrication methods (gaseous-phae, liquid-phase, solid-phase, template synthesis-concepts with examples only). Nano perspective of the fabrication methods.

### Unit II: Chemical Synthesis-I

i) Combustion: Chemical etching of silicon ii) Basic concepts of Chemical-Mechanical polishing.

iii) Anodization and Electropolishing: Chemical reactions of electrodeposition of aluminum.

### Unit III: Chemical Synthesis-II

Introduction to molecular self-assembly (MSA), Template synthesis, Sol-gel methods, metal reduction, emulsion polymerization, block copolymerization, electrodeposition with examples and reactions involved.

### **Unit IV: Biological Synthesis**

Biological synthesis of Nanoparticles, Concept of reducing and capping agents, introduction to biomolecules as reducing and capping agents, Bacteria, fungi and plants as sources of reducing and capping agents and for biogenic synthesis of nanomaterials. Advantages and applications of biologically synthesized nanomaterials. Introduction to biological nanomaterials. Biomineralization, Magnetosomes, DNA based Nano structures, Protein based Nano structures

#### **References:**

1. Introduction to nanoscience and nanotechnology, CRC Press, Tylor and Francis Group, Boca Raton, G. L. Hornyak, H. F. Tibbals, J. Dutta and J J. Moore.

2. Introductory Nanoscience: Physical and Chemical Concepts, CRC Press, Tylor and Francis Group, Boca Raton, M. Kuno.

3. Nanotechnology-S. K. Kulkarni (3<sup>rd</sup> Edition)

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### (B. Sc. Nanotechnology) F. Y. B. Sc., Sem. II Syllabus Paper II

### Title of Paper: Basic Characterization Techniques for Nanomaterials Subject Code: DSC-38B

### **Unit I:Introduction to Nanotools**

Types of characterization methods: Electron probe methods, Scaning Probe methods, Spectroscopic methods and Nonradiative and Nonelectron methods (classification and concepts only). Optics and resolution (formula and calculations).

### Unit II:Fundamentals of Spectroscopy I

Electromagnetic radiation and range (problems on interconversion of wavelength to frequency and energy). Relationship between electromagnetic radiation range and spectroscopy. Fundamentals and working principle of UV-Visible spectroscopy, difference between absorbance and surface plasmon resonance (SPR), principle of Fourier-Transformation, fundamentals and working principle of FT-IR, application in functional group determination of organic compounds (-OH, -COOH, -NH<sub>2</sub>, -NH-, -O-).

### Unit III: Fundamentals of Spectroscopy II

Emission spectroscopy: Fundamentals and working principle of spectrofluorometer, concept of singlet and triplet electronic state, definition of fluorescence and phosphorescence through Jablonski diagram. Distinction between radiative and nonradiative emissions. Definition of luminescence and basic types (chemiluminiscence).

#### **Unit IV: Tools in Nanobiotechnology**

Fundamentals of Microscopy, types of microscopes, light microscope, compound microscope, bright field and dark phase microscopy, inverted microscope, Confocal microscopy. Applications of microscopy in nanobiotechnology.

#### **References:**

1. Introduction to nanoscience and nanotechnology, CRC Press, Tylor and Francis Group, Boca Raton, G. L. Hornyak, H. F. Tibbals, J. Dutta and J J. Moore.

2. Introductory Nanoscience: Physical and Chemical Concepts, CRC Press, Tylor and Francis Group, Boca Raton, M. Kuno.

3. Nanotechnology-S. K. Kulkarni (3rd Edition)

4. Fundamentals of Molecular Spectroscopy, C. N. Banwell and E. M. McCash (4<sup>th</sup> Edition), Tata McGraw-Hill Publishing Company Ltd., New Delhi.

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### Title: Laboratory Course

# List of Experiments for B.Sc. Nanotechnology

# F.Y.B.Sc. (Sem I &II)

| Sr.No. | Name of the Experiments   |  |  |
|--------|---|--|--|
| 1.     | Reduction of $Ag^+$ by glucose (Tollen's Test for reducing sugar).  |  |  |
| 2.     | Reduction of Cu <sup>2+</sup> by aldehydes (Fehling's Test).  |  |  |
| 3.     | Synthesis of Ag nanoparticles using sodium borohydride (Creighton's method).  |  |  |
| 4.     | Synthesis of Au nanoparticles using citric acid (Lee – Meisel method)   |  |  |
| 5.     | Determination of density of colloidal Ag/Au nanoparticles using specific gravity bottle (5 ml).   |  |  |
| 6.     | Measurements of conductivity of KCl solution at different concentration.  |  |  |
| 7.     | Volumetric acid base titration using pH meter.(Strong acid vs Strong base , Weak acid vs Strong base )  |  |  |
| 8.     | Determination of viscosity of polymer by using viscometer.  |  |  |
| 9.     | Diffraction grating by LASER.   |  |  |
| 10.    | Determination of surface tension of a liquid by drop weight method.   |  |  |
| 11.    | Error analysis.   |  |  |
| 12.    | I-V characteristics of solar cells.   |  |  |
| 13.    | Demonstration of nano TiO <sub>2</sub> dye sensitized solar cell. – NVIS kit  |  |  |
| 14.    | Demonstration of nanotechnology. – NVIS kit   |  |  |
| 15.    | Use of multimeter to measure resistance/Inductance/diode/transistor.  |  |  |
| 16.    | Validation of Lambert's-Beer's law using CuSO <sub>4</sub> solution.  |  |  |
| 17.    | Calculation of total number of atom's and surface atom's present in a nano-<br>partical of a given size. (Theoretical)  |  |  |
| 18.    | Calculation of surface area to volume ratio of 1D solid and hollow nanostructure. (Theoretical).  |  |  |
| 19.    | Calculate the ballistic I-V characteristics for conventional MOSFETs, Nanowires MOSFETs and Carbon Nanotube MOSFETs (FETToy)  |  |  |
| 20.    | Introduction to Nanobiotechnology lab equipments- Calorimeter, pH meter,<br>Weighing balance, Hot Air Oven, Water Bath, Autoclave, Laminar Air Flow, -20<br>°C deep freezer, Thermal cycler machine (PCR machine), Gel Electrophoresis<br>system, PAGE system, power supply/unit, Centrifuge, water distillation unit etc |  |  |
| 21.    | Preparation of stock solutions and Buffer Solution, Stock Solution, such as<br>Acetate Buffer pH 4.8, phosphate buffer pH 9.6, Phosphate Buffer saline pH 7.2,<br>Saline solution pH 7.0  |  |  |
| 22.    | Preparation of Plant extract (Organic and aqueous), Crushing, grinding,<br>maceration, homogenization, Filtration, Centrifugation, cold percolation<br>extraction, hot extraction, using Sohxlet apparatus  |  |  |